

## Z-Selective Intramolecular Horner–Wadsworth–Emmons Reaction for the Synthesis of Macrocyclic Lactones

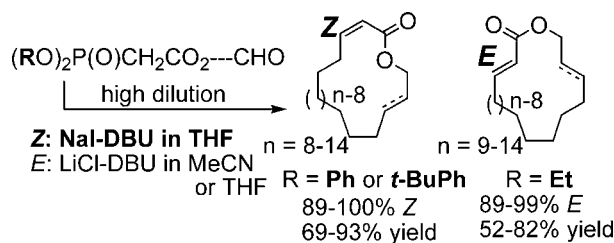
Kaori Ando,<sup>\*,†</sup> Kenji Narumiya,<sup>†</sup> Hirokazu Takada,<sup>†</sup> and Taiji Teruya<sup>‡</sup>

Department of Chemistry, Faculty of Engineering, Gifu University, Yanagido 1-1, Gifu 501-1193, Japan, and College of Education, University of the Ryukyus, Nishihara-cho, Okinawa 903-0213, Japan

ando@gifu-u.ac.jp

Received January 12, 2010

## ABSTRACT

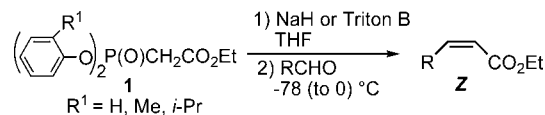


When the substrates  $(ArO)_2P(O)CH_2CO_2---CHO$  ( $Ar = Ph, o\text{-}t\text{-BuPh}$ ) were added to a THF solution containing 3 equiv of NaH at 0 °C or NaI–DBU at rt over 1–10 h, the intramolecular Horner–Wadsworth–Emmons reaction proceeded efficiently to give the 12–18-membered-ring lactones in 69–93% yields with 89–100% Z selectivity. On the other hand,  $(EtO)_2P(O)CH_2CO_2---CHO$  gave the 13–18-membered-ring lactones in 52–82% yields with 89–99% E selectivity using LiCl–DBU in MeCN or THF.

Since there are a variety of macrocyclic lactones, so-called macrolides in natural products, and they have an interesting biological activity, a number of methods for their construction have been reported. After the first reports by Stork and Nakamura<sup>1</sup> and Nicolaou et al.<sup>2</sup> in 1979, the intramolecular Horner–Wadsworth–Emmons (HWE) reaction has been one of the most reliable macrocyclization procedures. These reactions gave *E*- $\alpha,\beta$ -unsaturated lactones with high selectivity. On the other hand, there are many bioactive macrolides containing a *Z*- $\alpha,\beta$ -unsaturated lactone moiety such as macrolactin A,<sup>3</sup> dictyostatin,<sup>4</sup> phorbioxazoles,<sup>5</sup> and laulimalide.<sup>6</sup> In general, the HWE reaction preferentially gives

more stable *E*-olefins. In order to prepare *Z*-olefins, the special HWE reagents were invented by Still et al. ( $(CF_3CH_2O)_2P(O)CH_2CO_2Me$ )<sup>7</sup> and by us (**1** in Scheme 1).<sup>8</sup>

Scheme 1



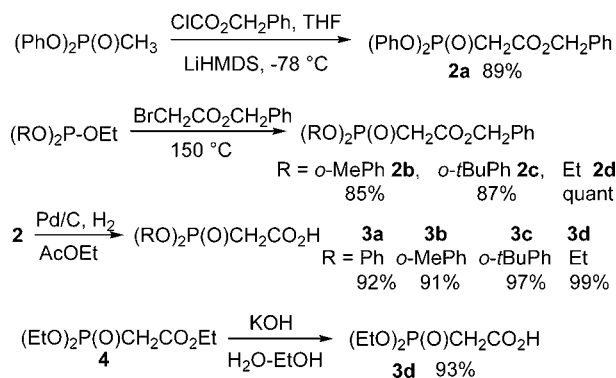
These reagents have been successfully applied for the synthesis of *Z*- $\alpha,\beta$ -unsaturated esters with high selectivity

<sup>†</sup> Gifu University.<sup>‡</sup> University of the Ryukyus.(1) Stork, G.; Nakamura, E. *J. Org. Chem.* **1979**, *44*, 4010–4011.(2) Nicolaou, K. C.; Seitz, S. P.; Pavia, M. R.; Petasis, N. A. *J. Org. Chem.* **1979**, *44*, 4011–4013.(3) Gustafson, K.; Roman, M.; Fenical, W. *J. Am. Chem. Soc.* **1989**, *111*, 7519–7524.(4) Pettit, G. R.; Cichacz, Z. A.; Gao, F.; Boyd, M. R.; Schmidt, J. M. *Chem. Commun.* **1994**, 1111–1112.(5) Searle, P. A.; Molinski, T. F. *J. Am. Chem. Soc.* **1995**, *117*, 8126–8131.(6) Quinoa, E.; Kakou, Y.; Crews, P. *J. Org. Chem.* **1988**, *53*, 3642–3644.(7) Still, W. C.; Gennari, C. *Tetrahedron Lett.* **1983**, *24*, 4405–4408.

from a variety of aldehydes. However, when these reagents were applied for the intramolecular HWE reaction, low to moderate selectivity and/or low yields were generally reported.<sup>9,10</sup> For example, Ghosh mainly received the *E* product for the synthesis of laulimalide.<sup>9a</sup> The *Z/E* ratios were 1:1.7 and 1:2 using our reagent and Still's reagent, respectively. After Aristoff's report,<sup>11,12</sup> K<sub>2</sub>CO<sub>3</sub> and 18-crown-6 in toluene have been used for the *Z*-selective intramolecular HWE reaction. In fact, all the intramolecular HWE reactions of ref 9 used these conditions. Since the reaction conditions for the *Z*-selective intramolecular HWE reaction have never been fully optimized, we decided to define the satisfactory reaction conditions. Here, we report our results of the intramolecular HWE reaction containing 12- to 18-membered lactones.

Benzyl diphenylphosphonoacetate **2a** was prepared by treatment of diphenyl methylphosphonate with 2 equiv of LiHMDS in the presence of ClCO<sub>2</sub>CH<sub>2</sub>Ph in THF, and the other phosphonates **2b–d** were obtained by heating (RO)<sub>2</sub>P(OEt)<sup>13</sup> with BrCH<sub>2</sub>CO<sub>2</sub>CH<sub>2</sub>Ph in high yields (Scheme 2). The phosphonoacetic acids **3a–d** were prepared by

Scheme 2

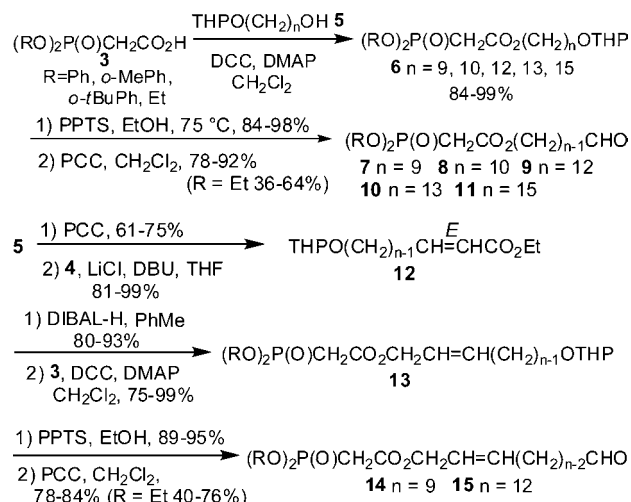


hydrogenolysis of **2** with Pd/C in 91–99% yields. Diethyl phosphonoacetic acid **3d** was also obtained by hydrolysis of **4** in 93% yield.

The intramolecular HWE substrates **7–11** were prepared by esterification of phosphonoacetic acid **3** with mono-THP-

protected diols **5**, followed by deprotection and oxidation as outlined in Scheme 3. Oxidation of **5** with PCC and the

Scheme 3



following HWE reaction with **4** using LiCl–DBU<sup>15,14</sup> gave *E*- $\alpha,\beta$ -unsaturated esters **12**. Reduction of **12** to the alcohols, esterification with **3**, deprotection, and then oxidation gave the intramolecular HWE substrates **14** and **15** containing the *E*-olefin. Although the yields were not optimized, the procedure is efficient especially for the diaryl phosphonoacetate reagents.

The intramolecular HWE reaction of **9** was summarized in Table 1.<sup>16</sup> The HWE reaction of **9a** was performed using a procedure similar to the intermolecular HWE reaction.<sup>8b</sup> That is, **9a** was treated with NaH in THF at –60 °C, and then the mixture was gradually warmed to 0 °C over 1 h. Only 12% yield of the 15-membered-ring lactone **Z-16** was obtained. The *Z/E* ratio of all the obtained HWE products is 92:8 (entry 1). The main products were oligomeric (dimeric and/or trimeric) lactones (58% yield). The formation of oligomers was suppressed when a THF solution of **9a** was added to a THF solution containing 3 equiv of NaH over 4 h at 0 °C (entry 2). Gladly and somewhat surprisingly, the *Z* selectivity did not reduce even at 0 °C, and a 73% yield of **Z-16** was obtained. **Z-16** was also efficiently obtained using NaI–DBU<sup>8c</sup> as a base in the highest yield (84%) (entry

(8) (a) Ando, K. *Tetrahedron Lett.* **1995**, *36*, 4105–4108. (b) Ando, K. *J. Org. Chem.* **1997**, *62*, 1934–1939. (c) Ando, K. *J. Org. Chem.* **1998**, *63*, 8411–8416. (d) Ando, K. *J. Org. Chem.* **1999**, *64*, 8406–8408. (e) Ando, K.; Oishi, T.; Hirama, M.; Ohno, H.; Ibuka, T. *J. Org. Chem.* **2000**, *65*, 4745–4749. (f) Ando, K. *J. Synth. Org. Chem., Jpn.* **2000**, *58*, 869–876.

(9) (a) Ghosh, A. K.; Wang, Y. *J. Am. Chem. Soc.* **2000**, *122*, 11027–11028. (b) Forsyth, C. J.; Ahmed, F.; Cink, R. D.; Lee, C. S. *J. Am. Chem. Soc.* **1998**, *120*, 5597–5598. (c) González, M. A.; Pattenden, G. *Angew. Chem., Int. Ed.* **2003**, *42*, 1255–1258. (d) Williams, D. R.; Kiryanov, A. A.; Emde, U.; Clark, M. P.; Berlinger, M. A.; Reeves, J. T. *Angew. Chem., Int. Ed.* **2003**, *42*, 1255–1258. Williams, D. R.; Clark, M. P. *Tetrahedron Lett.* **1999**, *40*, 2291–2294. (e) Ahmed, A.; Hoegenauer, E. K.; Enev, V. S.; Hanbauer, M.; Kaehlig, H.; Ohler, E.; Mulzer, J. *J. Org. Chem.* **2003**, *68*, 3026–3042. (f) Pattenden, G.; Gonzalez, M. A.; Little, P. B.; Millan, D. S.; Plowright, A. T.; Tornos, J. A.; Ye, T. *Org. Biomol. Chem.* **2003**, *1*, 4173–4208. (g) Smith, A. B., III; Minbiole, K. P.; Verhoest, P. R.; Schelhaas, M. *J. Am. Chem. Soc.* **2001**, *123*, 10942–10953. Smith, A. B., III; Razler, T. M.; Ciavarrri, J. P.; Hirose, T.; Ishikawa, T. *Org. Lett.* **2005**, *7*, 4399–4402. Smith, A. B., III; Razler, T. M.; Meis, R. M.; Pettit, G. R. *Org. Lett.* **2006**, *8*, 797–799. (h) O'Neil, G. W.; Phillips, A. J. *J. Am. Chem. Soc.* **2006**, *128*, 5340–5341.

(10) Zhang, J.; Xu, X. *Tetrahedron Lett.* **2000**, *41*, 941–943.

(11) Aristoff, P. A. *J. Org. Chem.* **1981**, *46*, 1954–1957.

(12) (a) Seo, S.-Y.; Jung, J.-K.; Paek, S.-M.; Lee, Y.-S.; Kim, S.-H.; Suh, Y.-G. *Tetrahedron Lett.* **2006**, *47*, 6527–6530. (b) Still, W. C.; Gennari, C.; Noguez, J. A.; Pearson, D. A. *J. Am. Chem. Soc.* **1984**, *106*, 260–262.

(13) Diaryl ethyl phosphites (ArO)<sub>2</sub>P(OEt) (Ar = *o*-MePh, *o*-*t*-BuPh) were prepared using Touchard's procedure.<sup>14</sup>

(14) Touchard, F. P.; Capelle, N.; Mercier, M. *Adv. Synth. Catal.* **2005**, *347*, 707–711.

(15) Blanchette, M. A.; Choy, W.; Davis, J. T.; Essensfeld, A. P.; Masamune, S.; Roush, W. R.; Sakai, T. *Tetrahedron Lett.* **1984**, *25*, 2183–2186.

(16) All the HWE products and the reagents described in this paper were characterized by 400 MHz <sup>1</sup>H NMR spectra and mass spectroscopy. The *Z/E* ratios were determined by integration of the vinyl proton signals.

**Table 1.** Intramolecular HWE Reaction of **9**

entry	<b>9</b>	solvent	base <sup>a</sup>	conditions <sup>b</sup>	yield of <b>Z-16</b> oligomers		Z/E
					(%)	(%)	
1	<b>9a</b>	THF	NaH	−60 to 0 °C <sup>c</sup>	12	58	92:8
2	<b>9a</b>	THF	NaH	0 °C, 4 h	73	1.2	92:8
3	<b>9a</b>	THF	NaI–DBU	0 °C, 4 h	<b>84</b>	2.3	<b>91:9</b>
4	<b>9a</b>	PhMe	K <sub>2</sub> CO <sub>3</sub> <sup>d</sup>	rt, 4 h	54	20	83:17
5	<b>9a</b>	THF	LiCl–DBU	rt, 10 h	74	<i>f</i>	81:19
6	<b>9b</b>	THF	NaI–DBU	0 °C, 10 h	70	3.5	97:3
7	<b>9b</b>	THF	NaI–DBU	rt, 10 h	<b>84</b>	0.2	<b>96:4</b>
8	<b>9b</b>	THF	NaH	0 °C, 10 h	<b>80</b>	0.2	<b>97:3</b>
9	<b>9b</b>	THF	NaH	rt, 10 h	60	0.1	96:4
10	<b>9c</b>	MeCN	LiCl–DBU	rt, 10 h	73 <sup>e</sup>	<i>f</i>	2:98

<sup>a</sup> 3 equiv of base was used except for entry 1 (1.4 equiv). <sup>b</sup> The addition was carried out over the specified time. The final concentration was 0.01 mol L<sup>−1</sup> except for entry 1 (0.05 mol L<sup>−1</sup>). <sup>c</sup> The reaction mixture was warmed over 1 h. <sup>d</sup> 18-crown-6 (5 equiv). <sup>e</sup> Yield of **E-16**. <sup>f</sup> Not detected.

3). Using the standard conditions for the *Z*-selective intramolecular HWE reaction, K<sub>2</sub>CO<sub>3</sub> and 18-crown-6 in toluene,<sup>11</sup> 54% yield of **Z-16** was obtained with 83:17 selectivity along with 20% oligomers (entry 4). In addition, LiCl–DBU in THF gave lower selectivity (81:19) (entry 5). Sodium base gave the highest *Z*-selectivity, the same as the intermolecular HWE reaction.<sup>8</sup> In our former study, we found that ortho-substituted phenyl reagents (R<sup>1</sup> = Me and *i*-Pr in **1**) showed higher *Z* selectivity. After that, Touchard et al. reported the improvement of selectivity at 0 °C using the *o*-*t*-BuPh reagent.<sup>14,17</sup> In the presence of 3 equiv of NaI–DBU, the *o*-*t*-BuPh reagent **9b** reacted to give **Z-16** in 70% yield with 97:3 selectivity at 0 °C. At room temperature, 84% yield was obtained with slightly reduced 96:4 selectivity (entry 7). The reaction using NaH gave similar results as NaI–DBU (entries 8–9). A better 80% yield was obtained at 0 °C with 97:3 selectivity. Thus, from the Ph reagent **9a** or *o*-*t*-BuPh reagent **9b**, **Z-16** was obtained in as high as 84% yield with 91–97% *Z* selectivity (entries 3, 7, and 8). On the other hand, the Et reagent **9c** was added to a solution containing 3 equiv of LiCl and DBU in MeCN to give **E-16** in 73% yield with 2:98 *E* selectivity (entry 10).

The results of the intramolecular HWE reaction of **8** were summarized in Table 2. Using both high dilution conditions and NaH as a base, the 13-membered-ring lactone **Z-17** was obtained in 52% yield with 94:6 selectivity from the Ph reagent **8a** (entry 1). Although the selectivity is lower, the highest yield of **Z-17** (75%) was obtained by using NaI–DBU at room temperature (entry 3). Disappointingly, *o*-MePh reagent **8b** gave lower 86:14 selectivity (entry 4). In the intermolecular HWE reaction, *o*-MePh reagent **1** (R<sup>1</sup> = Me) gave higher *Z*-selectivity (93–99% for a variety of alde-

**Table 2.** Intramolecular HWE Reaction of **8**

entry	<b>8</b>	base <sup>a</sup>	conditions <sup>b</sup>	yield of <b>Z-17</b> oligomers		Z/E
				(%)	(%)	
1	<b>8a</b>	NaH	0 °C, 4 h	52	17	94:6
2	<b>8a</b>	NaI–DBU	0 °C, 4 h	69	8	90:10
3	<b>8a</b>	NaI–DBU	rt, 4 h	<b>75</b>	7	<b>89:11</b>
4	<b>8b</b>	NaI–DBU	rt, 10 h	63	16	86:14
5	<b>8c</b>	NaI–DBU	rt, 10 h	<b>69</b>	4	<b>93:7</b>
6	<b>8c</b>	NaH	rt, 10 h	59	3	99:1
7	<b>8c</b>	NaH	0 °C, 10 h	<b>60</b>	7	<b>99:1</b>
8	<b>8d</b>	LiBr–DBU	rt, 4 h	52 <sup>d</sup>	12	6:94
9 <sup>1</sup>	<b>8d</b>	LiHMDS <sup>c</sup>	rt, 15 h	66 <sup>d</sup>	<1	0:100

<sup>a</sup> 3 equiv of base was used. <sup>b</sup> The addition was carried out over the specified time. The final concentration was 0.01 mol L<sup>−1</sup>. <sup>c</sup> 1.1 equiv base in the presence of 1% HMPA. <sup>d</sup> Yield of **E-17**.

hydes) than the Ph reagent at −78 °C or by warming from −78 to 0 °C.<sup>8b</sup> The *o*-*t*-BuPh reagent **8c** gave **Z-17** in 69% yield with 93:7 selectivity using NaI–DBU and in 60% yield with 99:1 selectivity using NaH (entries 5 and 7). Generally speaking, NaH gave the same or slightly higher *Z* selectivity than NaI–DBU. The yields were higher at 0 °C for NaH and at room temperature for NaI–DBU. Since the formation of the smaller 13-membered ring is rather difficult, the Ph reagent **8a** may be the reagent of choice because of its higher reactivity. The Et reagent **8d** gave **E-17** in 52% yield using LiBr–DBU in THF together with oligomers (12% yield) (entry 8). Although we did not optimize the *E*-selective reaction, slower addition might increase the yield of **E-17**. Stork and Nakamura reported the exclusive synthesis of **E-17** in 66% yield by the addition of **8d** over 15 h.<sup>1</sup>

Similarly, a solution of **10** (R = *o*-*t*-BuPh) was added to a THF solution containing 3 equiv of NaI and DBU at room temperature over 10 h to give the 16 membered ring lactone **Z-18** in 93% yield (Z/E = 94:6) (Figure 1). The addition time can be reduced for easier cases. In fact, **Z-18** was obtained in 89% yield (Z/E = 95:5) by the addition of **10** over 1 h. Using NaH at 0 °C gave **Z-18** in 76% yield with 94:6 *Z* selectivity. While, the reaction of **10** (R = Et) using 3 equiv of LiCl and DBU in MeCN gave **E-18** in 82% yield with 2:98 *E*-selectivity. The 18-membered-ring lactone **Z-19** was obtained in 71% yield (Z/E = 96:4) using NaH at 0 °C and in 77% yield (Z/E = 94:6) using NaI–DBU at room temperature from **11** (R = *o*-*t*-BuPh). On the other hand, **11** (R = Et) gave **E-19** in 82% yield (Z/E = 2:98). In addition, the 12-membered-ring lactone **Z-20** was obtained in 69% yield (Z/E = 99:1) and 67% yield (Z/E = 100:0) by the addition of **7** (R = *o*-*t*-BuPh) over 10 and 1 h, respectively. Interestingly, the *Z* selectivity obtained from 1 h addition was slightly better than the result from 10 h addition for

(17) For the amide reagents, (*o*-*t*-BuPhO)<sub>2</sub>P(O)CH<sub>2</sub>CONR<sup>1</sup>R<sup>2</sup>, see: Ando, K.; Nagaya, S.; Tarumi, Y. *Tetrahedron Lett.* **2009**, *50*, 5689–5691.

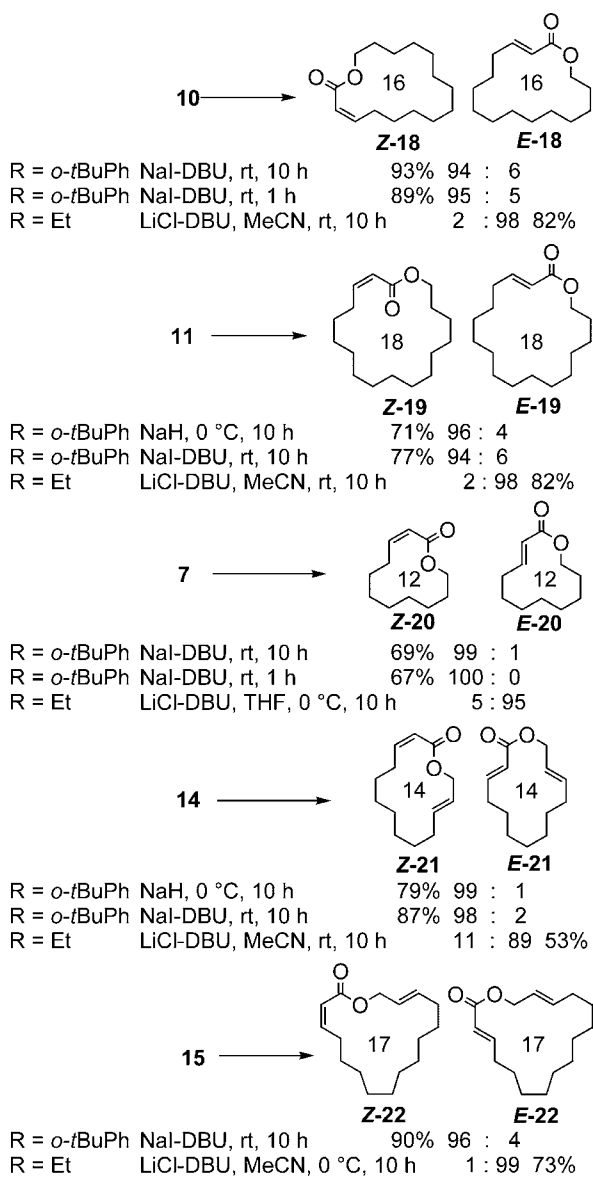


Figure 1

both **7** and **10**. Using NaH at 0 °C gave **Z-20** in 55% yield with 100% *Z* selectivity. The HWE reaction of **7** (R = Et) gave the HWE products E1, E2, E3, and Z1 in 92% combined yield (43:32:20:5 ratio), and the *Z/E* ratio of all the obtained HWE products was 5:95. Since these HWE products were not perfectly separated, the monomer **E-20** has not been specified yet. The 14-membered-ring lactone **Z-21** having an *E*-alkenyl group was prepared from **14** (R = *o*-*t*-BuPh) using either NaH at 0 °C or NaI-DBU at room temperature in 79 or 87% yield highly selectively (98–99% *Z*), while **14** (R = Et) gave **E-21** in 53% yield with 11:89 *E* selectivity. The 17-membered-ring lactone **Z-22** was also prepared from **15** (R = *o*-*t*-BuPh) using NaI-DBU in 90% yield, and **E-22** was obtained from **15** (R = Et) in 73% yield with 1:99 *E* selectivity.

In summary, the intramolecular HWE reaction of the substrates (ArO)<sub>2</sub>P(O)CH<sub>2</sub>CO<sub>2</sub>---CHO (Ar = Ph, *o*-*t*-BuPh) gives the 12–18-membered-ring *Z*-lactones in good to high yields using NaI-DBU at room temperature or NaH at 0 °C in THF. The substrate should be added to a THF solution containing 3 equiv of base over 1–10 h. The 13–18-membered *E*-lactones were also selectively obtained from (EtO)<sub>2</sub>P(O)CH<sub>2</sub>CO<sub>2</sub>---CHO. In this study, we succeeded in showing that the *Z*-selective intramolecular HWE reaction is very useful for the synthesis of simple *Z*- $\alpha,\beta$ -unsaturated lactones. As Stork mentioned, the most difficult cyclization will, in general, involve the least-substituted precursors because of more degrees of freedom.<sup>1</sup> We are planning to use this method for the synthesis of naturally occurring bioactive macrolides containing *Z*- $\alpha,\beta$ -unsaturated lactone moiety.

**Acknowledgment.** This work was partially supported by a Grant-in-Aid for Scientific Research from the Ministry of Education, Culture, Sports, Science, and Technology, Japan.

**Supporting Information Available:** A typical experimental procedure and compound characterization data. This material is available free of charge via the Internet at <http://pubs.acs.org>.

OL100071D